

Asymmetric Synthesis of the Enantiomers of the Diarylcarbinol (1*R*)- and (1*S*)-1-(1-Hydroxyphenylmethyl)-2-hydroxybenzene

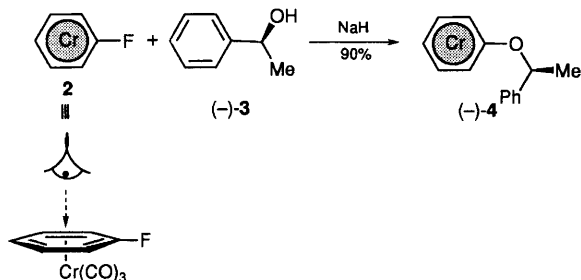
Stephen G. Davies* and W. Ewan Hume

The Dyson Perrins Laboratory, University of Oxford, South Parks Road, Oxford, UK OX1 3QY

(1*S*)-1-Phenethyl(phenyl chromium tricarbonyl) ether (–)-**4** undergoes completely stereoselective *ortho*-deprotonation with the thus formed carbanion being converted by the complementary sequences, benzoylation/hydride reduction and formylation/PhMgBr addition, to generate after decomplexation and deprotection the homochiral diarylcarbinols (–)-(1*R*)- and (+)-(1*S*)-1-(1-hydroxyphenylmethyl)-2-hydroxybenzene **1**, respectively.

Attachment of a chromium tricarbonyl unit to an arene changes its chemical characteristics.¹ The ring proton acidities are enhanced and the complexed arene becomes susceptible to nucleophilic attack.² Furthermore 1,2-differentially substituted arene chromium tricarbonyl complexes are chiral and such complexes have been resolved by a variety of methods including the separation of diastereoisomers³ and kinetic resolutions.⁴ In this area, attention has recently been focused on the asymmetric synthesis of arene chromium tricarbonyl complexes *via* the diastereoselective *ortho*-substitution of a phenyl chromium tricarbonyl moiety attached to a chiral auxiliary,⁵ and the *ortho*-deprotonation of an acetal derived from benzaldehyde chromium tricarbonyl by a homochiral lithium amide base.⁶ We report herein a new auxiliary, derived from 1-phenethanol, for the diastereoselective *ortho*-lithiation of a complexed phenyl ring, which results in the highest diastereoselectivities to date, and its use in the asymmetric synthesis of the homochiral diarylcarbinols (1*R*)- and (1*S*)-1-(1-hydroxyphenylmethyl)-2-hydroxybenzene **1**. Diarylcarbinols are an important class of compounds⁷ difficult to obtain in homochiral form by other methods.

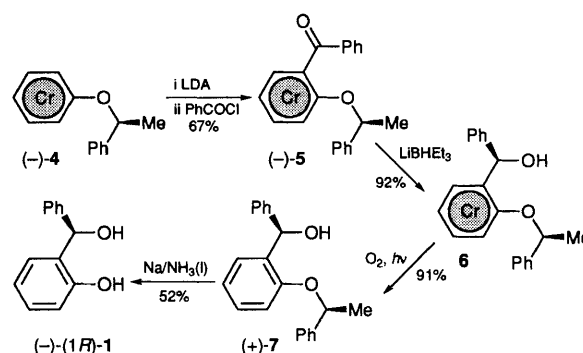
The auxiliary 1-phenethyl was introduced *via* an *ipso*-substitution reaction between tricarbonyl (fluorobenzene) chromium **2** and phenethanol **3**. Treatment of **2** with the sodium salt of (–)-(1*S*)-1-phenethanol **3**, resulted in the formation of (1*S*)-1-phenethyl(phenyl chromium tricarbonyl) ether (–)-**4** $\{[\alpha]_D^{24} -304.9$ (*c* 0.30, CH₂Cl₂) $\}$ as a yellow crystalline material. The diastereotopic nature of the *ortho* protons in (–)-**4** was evidenced by their chemical shifts at δ_H 4.86 and 5.27.



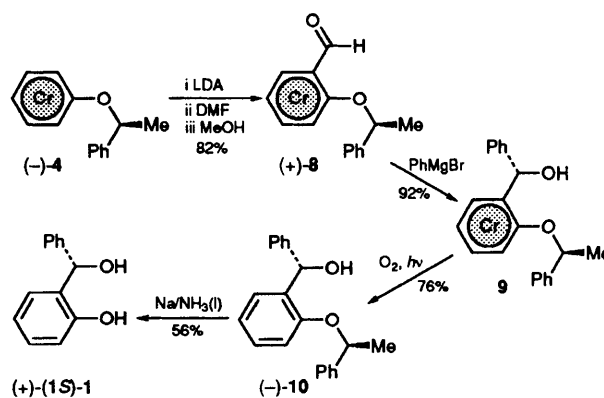
Deprotonation of (–)-**4** with lithium diisopropylamide (LDA, 1.5 equiv., –78 °C, THF) followed by quenching of the resultant anion with benzoyl chloride (5 equiv.) gave the red crystalline complex (–)-**5** $\{[\alpha]_D^{25} -200.0$ (*c* 0.26, CH₂Cl₂) $\}$ as a single diastereoisomer. ¹H NMR spectroscopic analysis of crude product **5** before purification was consistent with completely regio- and diastereo-selective deprotonation of **4** by LDA.

It has been established previously that *ortho*-substituted aldehydes and ketones undergo highly stereoselective addition reactions with nucleophiles approaching the unhindered face of the carbonyl, *anti* to the chromium tricarbonyl moiety, in the *anti* C=O to *ortho*-substituent, conformation.^{4a,8} Thus superhydride[®] reduction (3.0 equiv., –78 °C, THF) of (–)-**5** gave the diarylcarbinol complex **6** completely stereoselectively. In the racemic series an X-ray crystal structure analysis established the relative configurations within (±)-**6** while the absolute config-

uration of **6** is assigned unambiguously from that of the starting (1*S*)-1-phenethanol. All attempts at removing the auxiliary while retaining the chromium tricarbonyl unit failed [H₂, 8 atm., Pd–C and Na/NH₃(l)] and therefore removal of the chromium tricarbonyl unit preceded removal of the auxiliary. Decomplexation (O₂, *hν*, Et₂O), generated (+)-**7** $\{[\alpha]_D^{25} +153.6$ (*c* = 0.13, CH₂Cl₂) $\}$ and removal of the auxiliary [Na/NH₃(l), –78 °C, 52%] gave (–)-1-[(1*R*)-1-hydroxyphenylmethyl]-2-hydroxybenzene **1** $\{[\alpha]_D^{25} -35.13$ (*c* 0.60, CH₂Cl₂) $\}$ which was confirmed to be homochiral (>98% ee) by ¹H NMR spectroscopic analysis upon comparison with authentic racemic (±)-**1** using (–)-(1*R*)-1-(9-anthryl)-2,2,2-trifluoroethanol as the chiral shift reagent.



Synthesis of (+)-(1*S*)-1-(1-hydroxyphenylmethyl)-2-hydroxybenzene **1** was then pursued, while still retaining (–)-(1*S*)-1-phenethanol as the source of the chiral auxiliary, *via* the complexed aldehyde (+)-**8**. Thus lithiation of (–)-**4** was carried out as before but quenched by the addition of dimethylformamide (DMF, 5.0 equiv.) to give (+)-**8** $\{[\alpha]_D^{24} +641.7$ (*c* 0.76, CH₂Cl₂) $\}$ as a single diastereoisomer. Addition of PhMgBr (4.0 equiv., –78 °C, THF) to the aldehyde complex (+)-**8** proceeded completely stereoselectively to give the secondary alcohol **9** (92%), the epimer of **6**, as a single diastereoisomer, *vide supra*. Removal of the chromium tricarbonyl unit generated (–)-**10** $\{[\alpha]_D^{25} -225.7$ (*c* 0.21, CH₂Cl₂) $\}$ the epimer of (+)-**7** while subsequent removal of the



auxiliary, as before, gave homochiral (>98% ee) (+)-(1*S*)-**1** {[α]_D²⁶ +36.1 (c 1.22, CH₂Cl₂)}.

Finally, deprotonation of **4** with LDA and quenching with benzaldehyde generated a 1:2 mixture of **6** and **9** indicating little selectivity in the formation of the benzylic stereogenic centre in this direct process.

In conclusion, the *O*-phenethyl chiral auxiliary in the complex **4** induces a completely stereoselective *ortho*-deprotonation reaction with LDA. The thus derived ketone **5** and aldehyde **8** undergo completely stereoselective hydride and Grignard additions respectively to generate, in a complementary fashion, following decomplexation and deprotection the homochiral diarylcarbinols (1*R*)- and (1*S*)-1-(1-hydroxyphenylmethyl)-2-hydroxybenzene **1**, respectively.

We thank the SERC for an earmarked studentship to W. E. H.

Received, 3rd November 1994; Com. 4/06729K

References

- 1 S. G. Davies, *Organotransition Metal Chemistry: Applications to Organic Synthesis*, Pergamon, Oxford, 1982.
- 2 S. G. Davies, *J. Organomet. Chem.*, 1990, **400**, 223; S. G. Davies and T. J. Donohoe, *Synlett*, 1993, 323.
- 3 A. Solladié-Cavallo, G. Solladié and E. Tsamo, *J. Org. Chem.*, 1979, **44**, 4189.
- 4 (a) S. G. Davies and C. L. Goodfellow, *J. Chem. Soc., Perkin Trans. I*, 1990, 393; (b) C. Baldoli, J. Gillois, G. Jaouen, S. Maiorana and S. Top, *J. Chem. Soc., Chem. Commun.*, 1988, 1284.
- 5 J. Blagg, S. G. Davies, C. L. Goodfellow and K. H. Sutton, *J. Chem. Soc., Perkin Trans. I*, 1987, 1805; S. J. Coote, S. G. Davies, C. L. Goodfellow, K. H. Sutton, D. Middlemiss and A. Naylor, *Tetrahedron Asymmetry*, 1990, **1**, 817; J. R. Green, J. Ho and Y. Kondo, *J. Org. Chem.*, 1991, **56**, 7199; 1993, **58**, 6182; J. Aubé, J. A. Heppert, M. L. Milligan, M. J. Smith and P. Zenk, *J. Org. Chem.*, 1992, **57**, 3563; M. Uemura, R. Miyake, K. Nakayama, M. Shiro and Y. Hayashi, *J. Org. Chem.*, 1993, **58**, 1238; P. W. N. Christian, R. Gil, K. Muniz-Fernandez, S. E. Thomas and A. Wierzchelyski, *J. Chem. Soc., Chem. Commun.*, 1994, 1569.
- 6 D. A. Price, N. S. Simpkins, A. M. MacLeod and A. P. Watt, *J. Org. Chem.*, 1994, **59**, 1961.
- 7 J. B. Chazan, F. Bellamy, D. Horton, J. Millet, F. Picart and S. Samreth, *J. Med. Chem.*, 1993, **36**, 898; L. Brahce, V. E. Gregor, S. J. Lobbestad, M. R. Pavia, D. R. Mayhugh, D. Nugiel, R. D. Schwarz, C. O. Taylor and M. G. Vartanian, *J. Med. Chem.*, 1992, **35**, 4238.
- 8 Y. Hayashi, K. Isobe, T. Kobayashi, T. Minami and M. Uemura, *J. Org. Chem.*, 1986, **51**, 2859.